

# Low Temperature Thermal Energy Storage

Insights into odd-even n-alkane systems

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# Contents

1

• OVERVIEW

3

• *N*-ALKANES

2

• EXPERIMENTAL WORK

4

• CONCLUSIONS

# Overview

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PCM systems  
for TES at low T

Cold Storage  
and  
Transportation

Phase  
Equilibrium  
Studies

$n$ -Alkanes  
 $n$ -alkyl adipates

4 different  
alkane  
systems

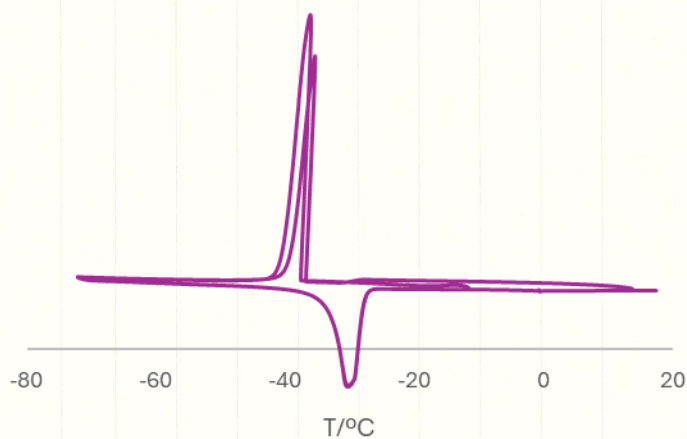
Odd and Even  
alkyl chains

Good potential  
as PCMs for  
low T

# Experimental Work

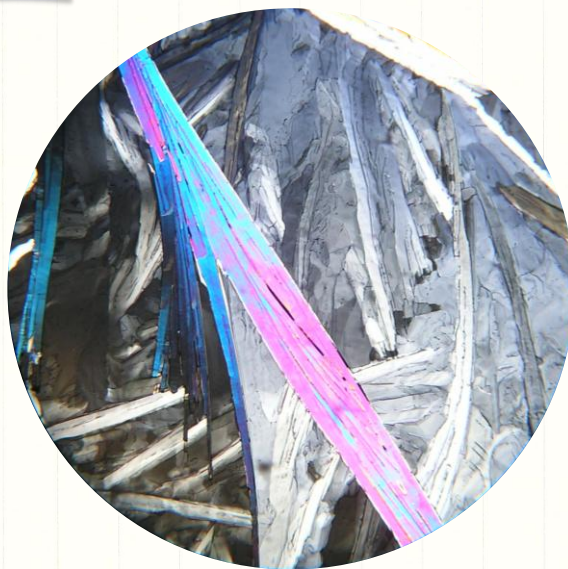
1

Differential Scanning Calorimetry (DSC)



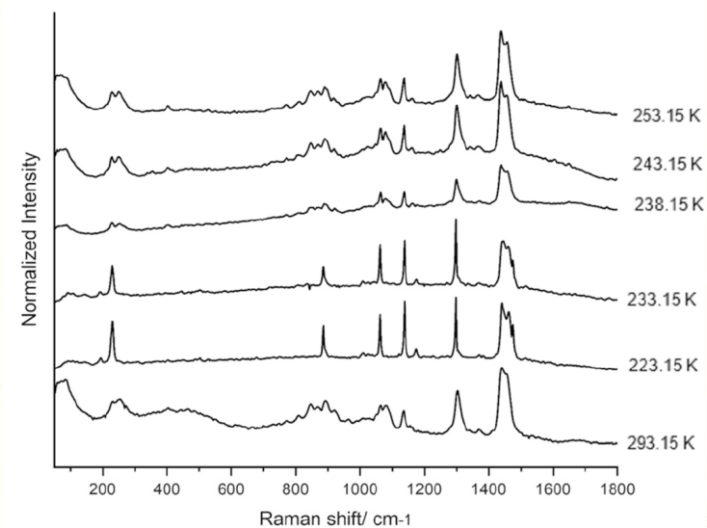
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Hot Stage Microscopy (HSM)



3

Raman Spectroscopy



# 1

# Differential Scanning Calorimetry (DSC)

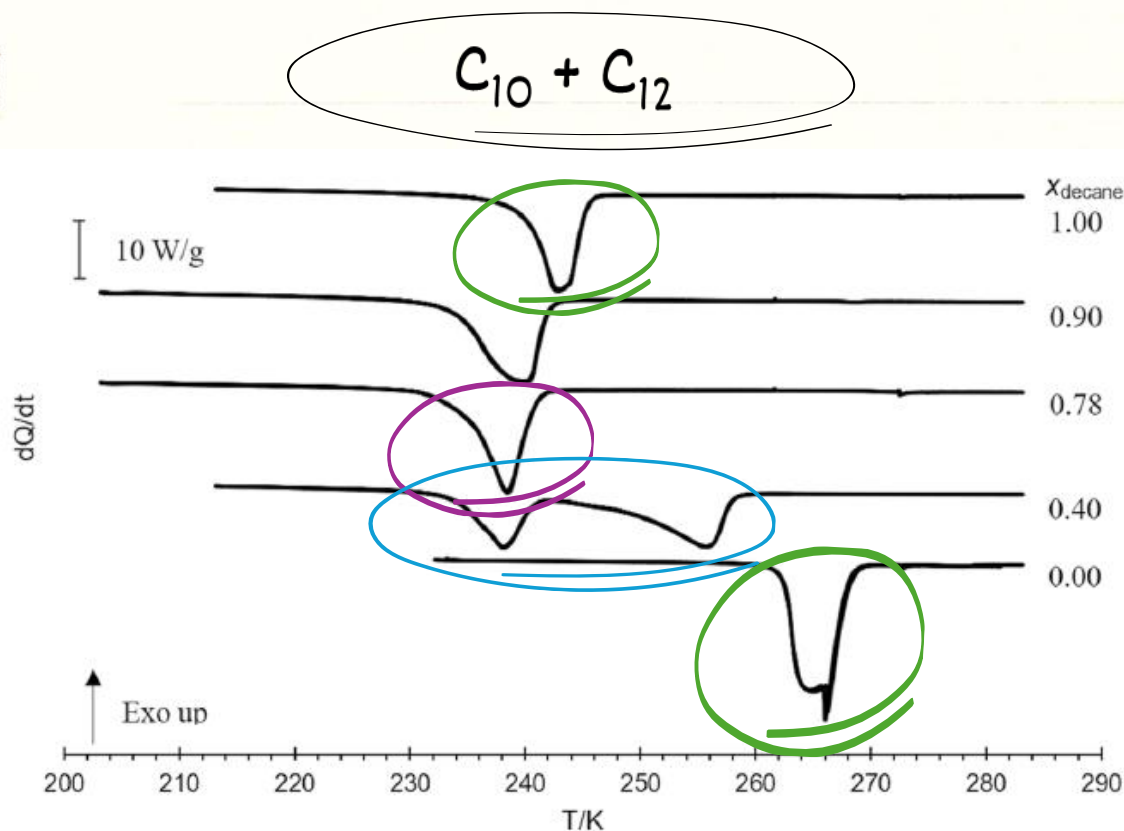


Fig. 1 – DSC heating curves of pure *n*-decane, *n*-dodecane, and of selected binary mixtures, with octane molar fraction  $x_{\text{decane}}$ \*

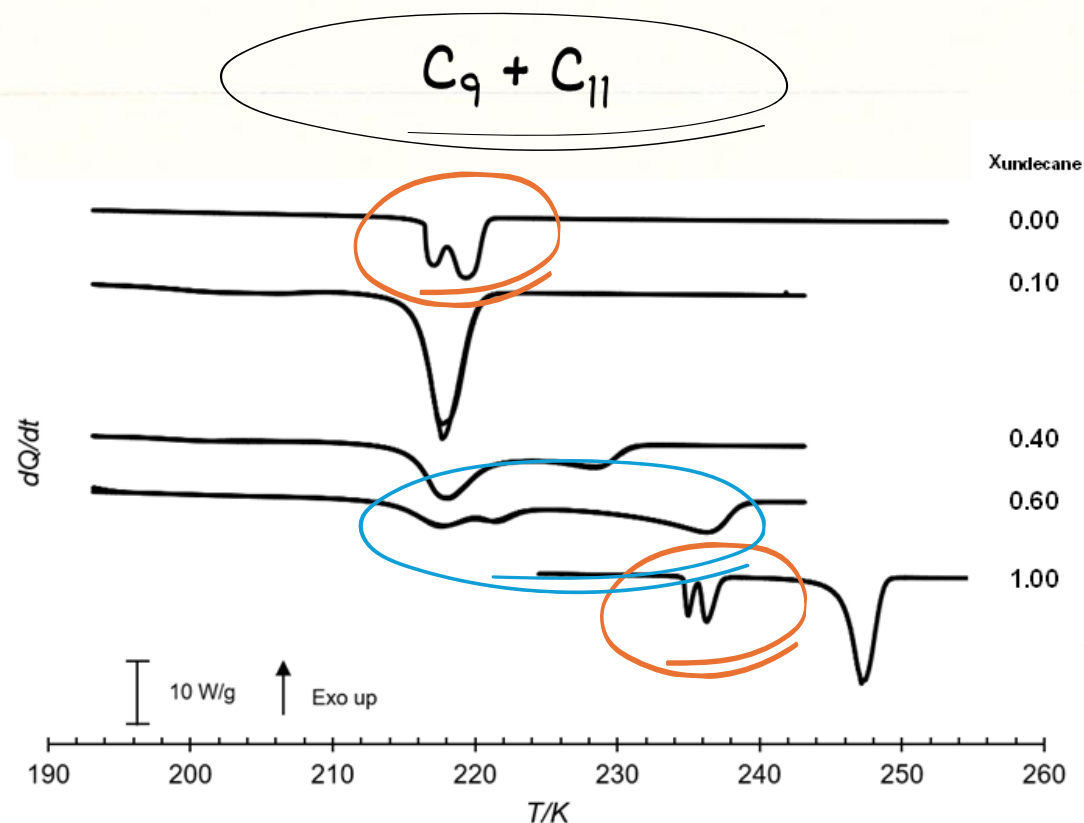


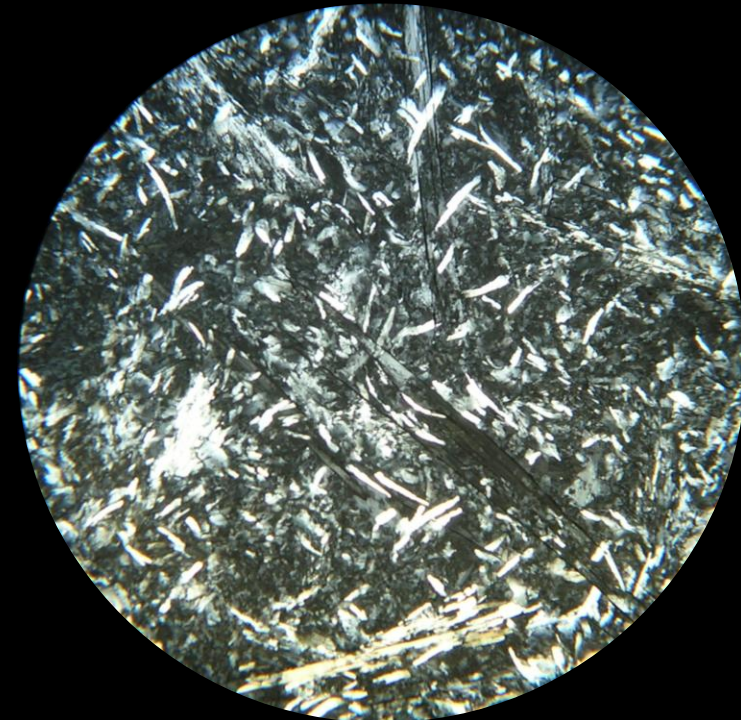
Fig. 2 – DSC heating curves of pure *n*-nonane, *n*-undecane, and of selected binary mixtures, with undecane molar fraction  $x_{\text{undecane}}$ \*

2

# Hot Stage Microscopy (HSM)



193.15 K



215.65 K

Fig. 3 – HSM images for the polymorphic transition of pure  $n\text{-C}_9$ .

2

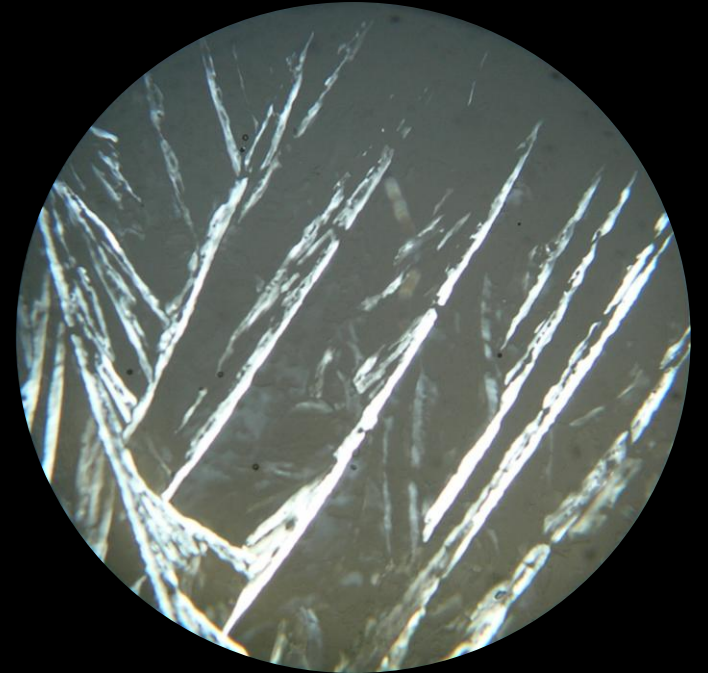
# Hot Stage Microscopy (HSM)



193.15 K



240.15 K



245.15 K

Fig. 4 – HSM images for the phase transition of pure  $n\text{-C}_{11}$ .

# 3

# Raman Spectroscopy

Pure compounds

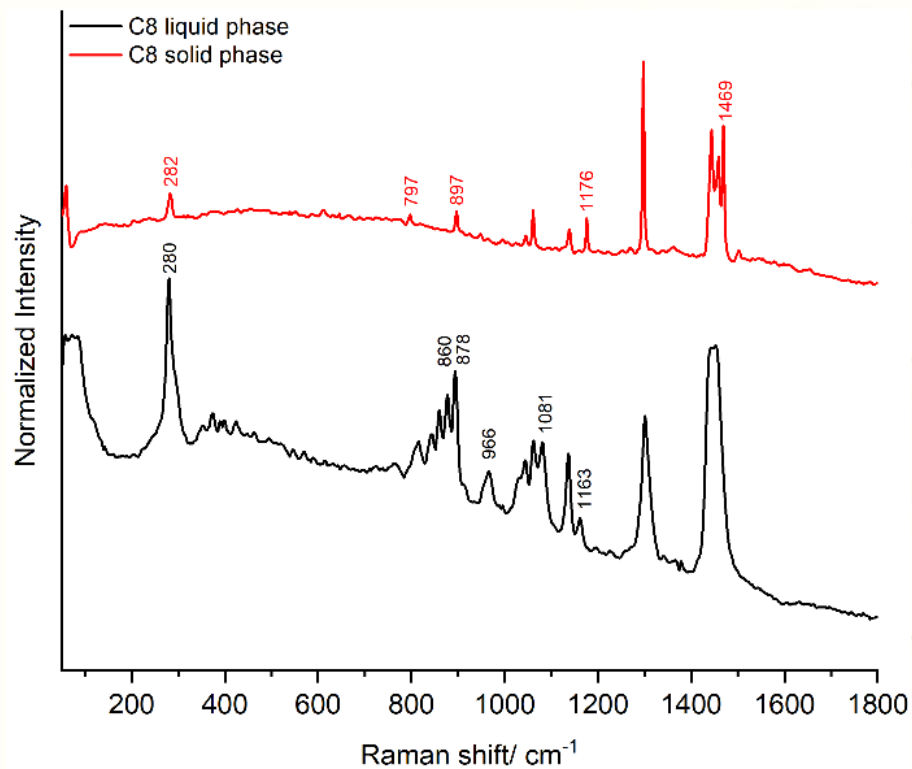


Fig. 5 – Raman spectra of solid and liquid phases samples of  $n\text{-C}_8$

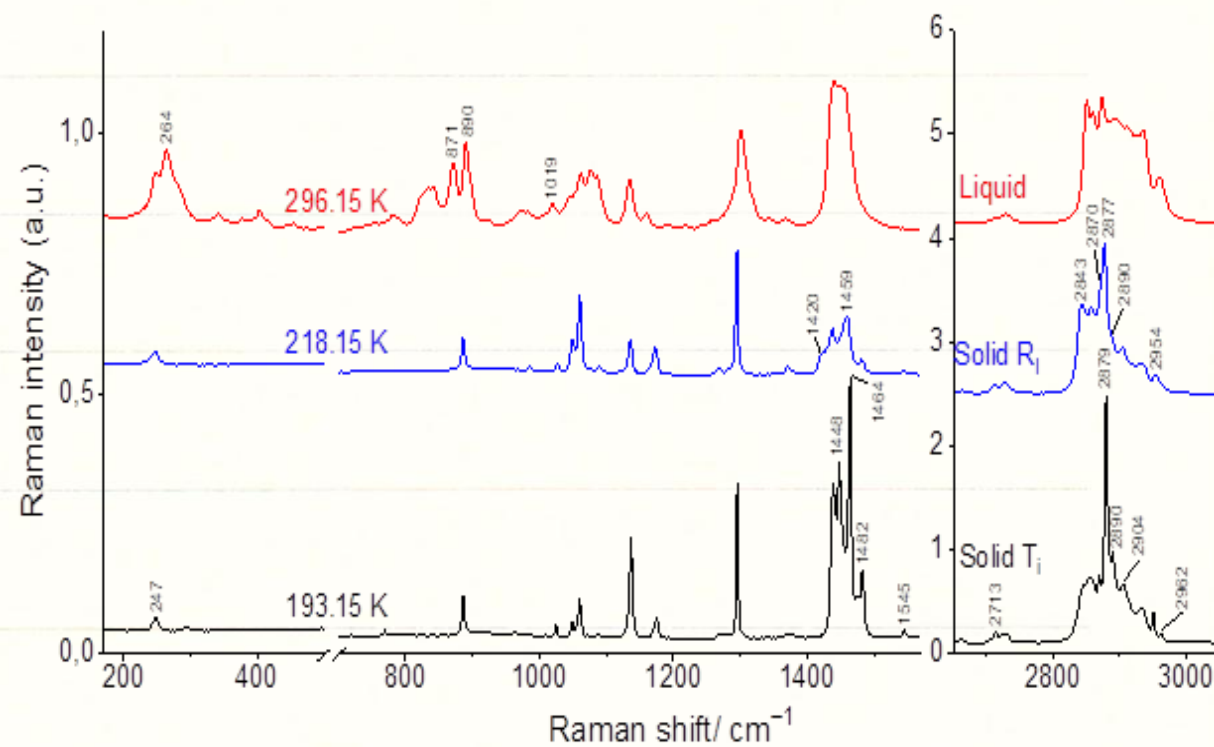


Fig. 6 – Raman spectra of solid and liquid phases samples of  $n\text{-C}_9$

## 3

## Raman Spectroscopy

## Binary Mixtures

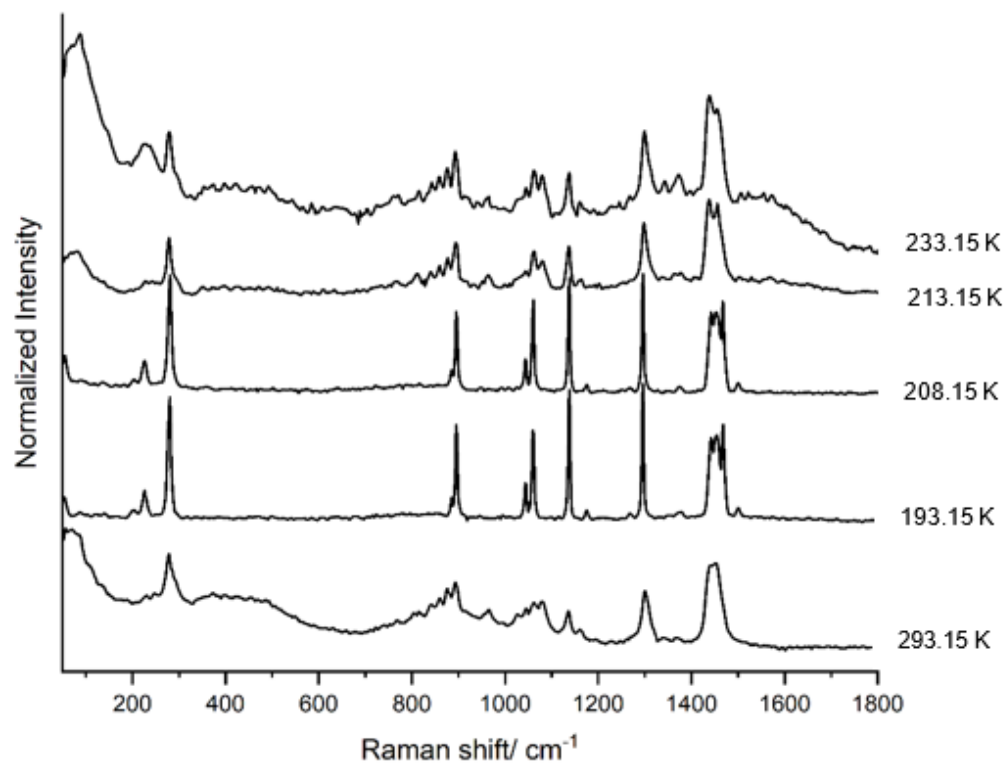


Fig. 7 - Temperature-variation Raman spectra for the eutectic mixture of the  $C_8/C_{10}$  binary system.

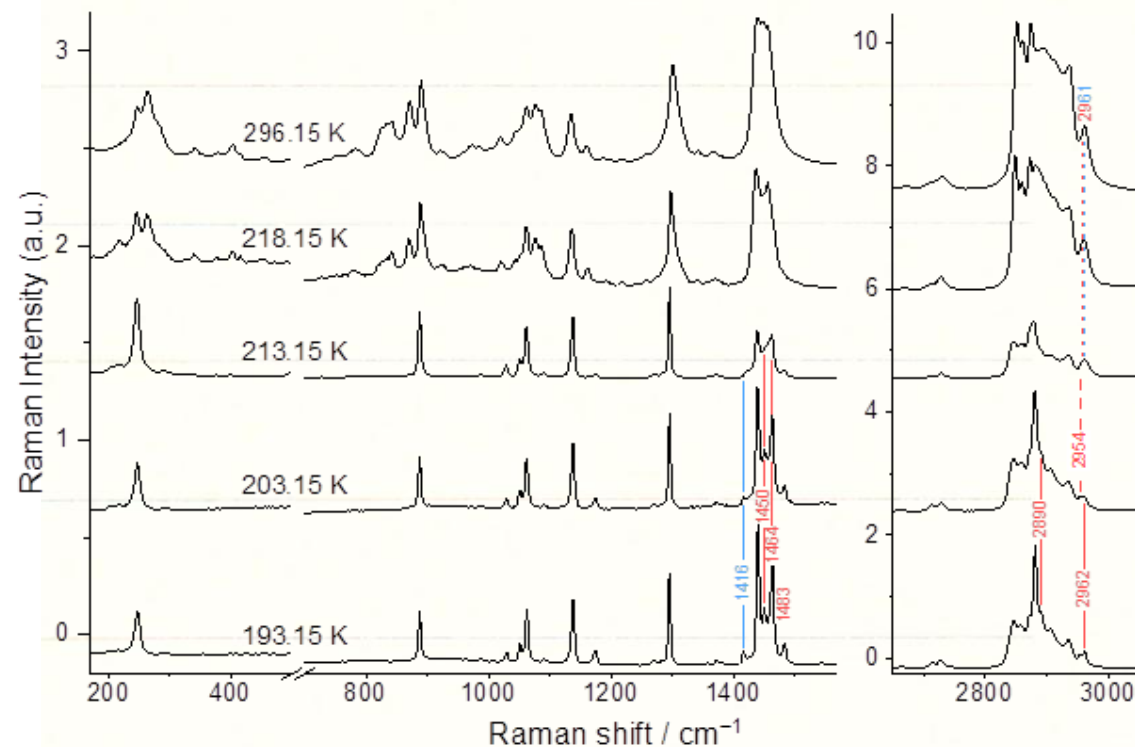
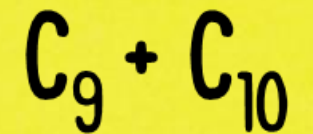
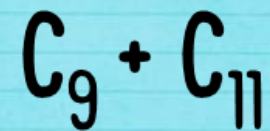
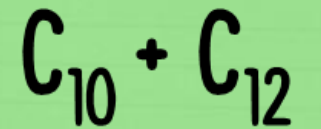
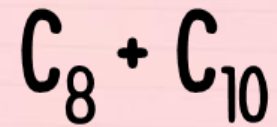


Fig. 8 - Temperature-variation Raman spectra for the congruent melting mixture of the  $C_9/C_{11}$  binary system.

# Binary Phase Diagrams



# Binary Phase Diagrams

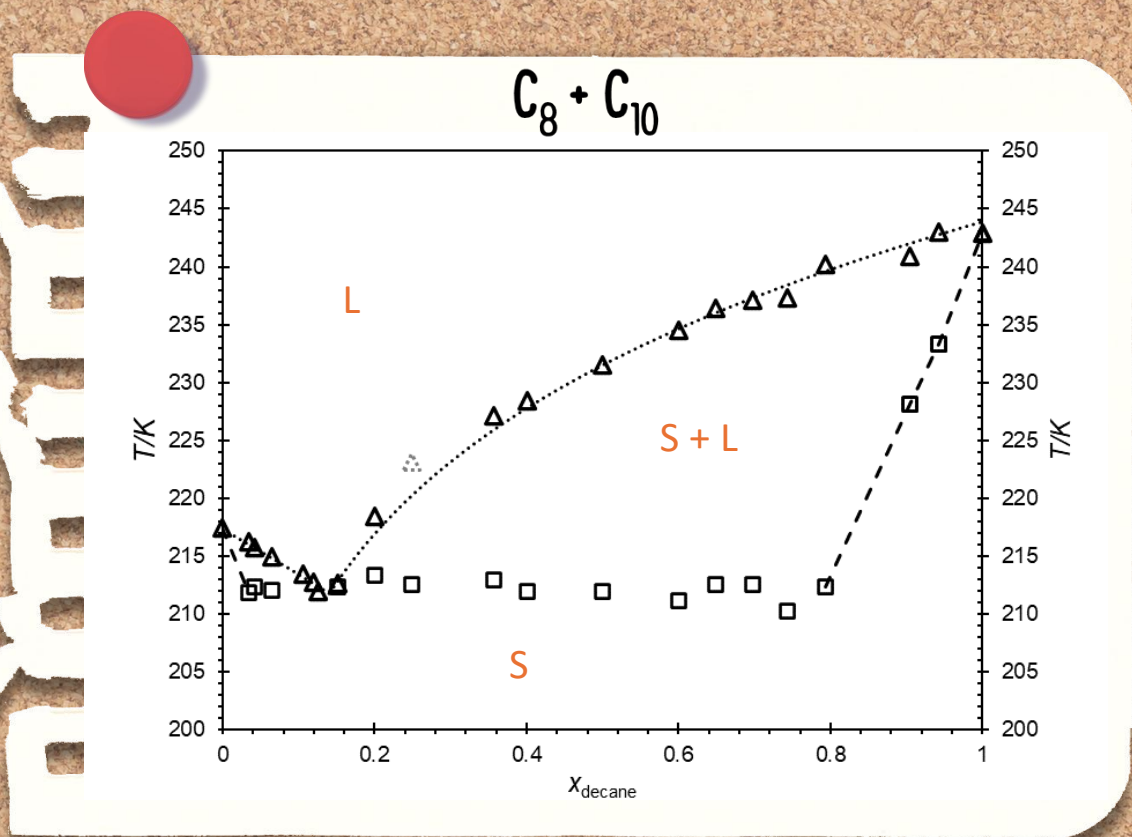


Fig. 9 - Binary solid-liquid phase diagram of *n*-C<sub>8</sub> and *n*-C<sub>10</sub>.

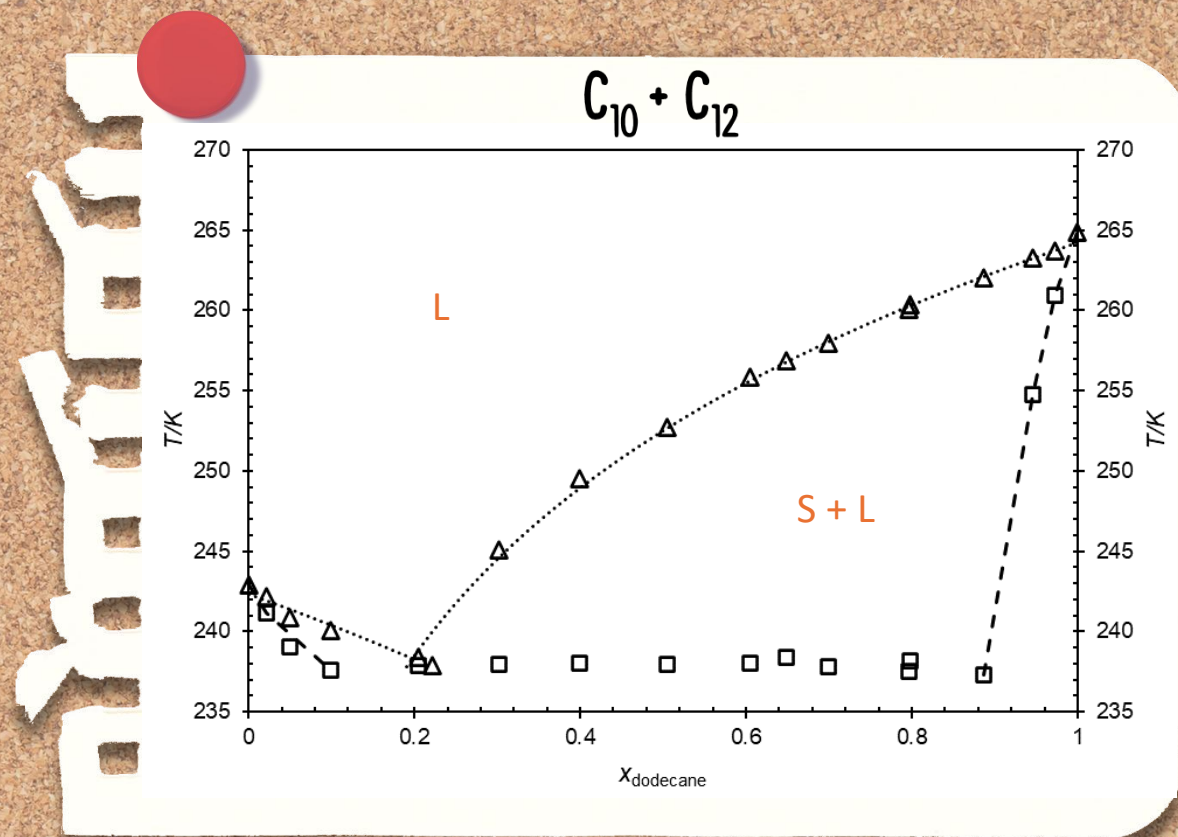


Fig. 10 - Binary solid-liquid phase diagram of *n*-C<sub>10</sub> and *n*-C<sub>11</sub>.

Results published in the International Journal of Thermophysics. DOI: 10.1007/s10765-023-03317-9

# Binary Phase Diagrams

$C_8 + C_{10}$

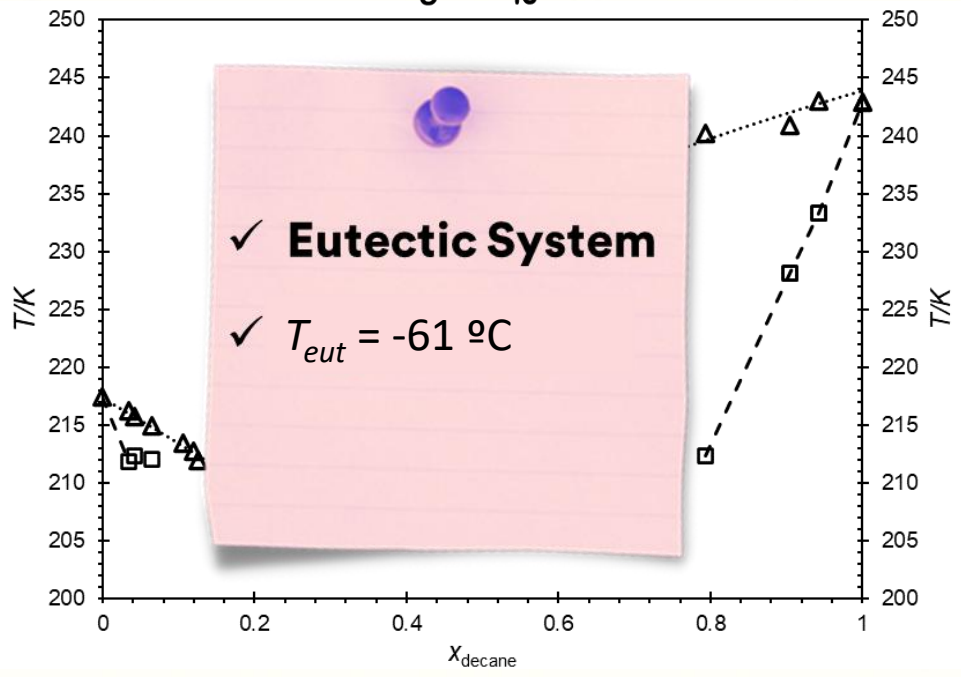


Fig. 9 - Binary solid-liquid phase diagram of  $n\text{-C}_8$  and  $n\text{-C}_{10}$ .

$C_{10} + C_{12}$

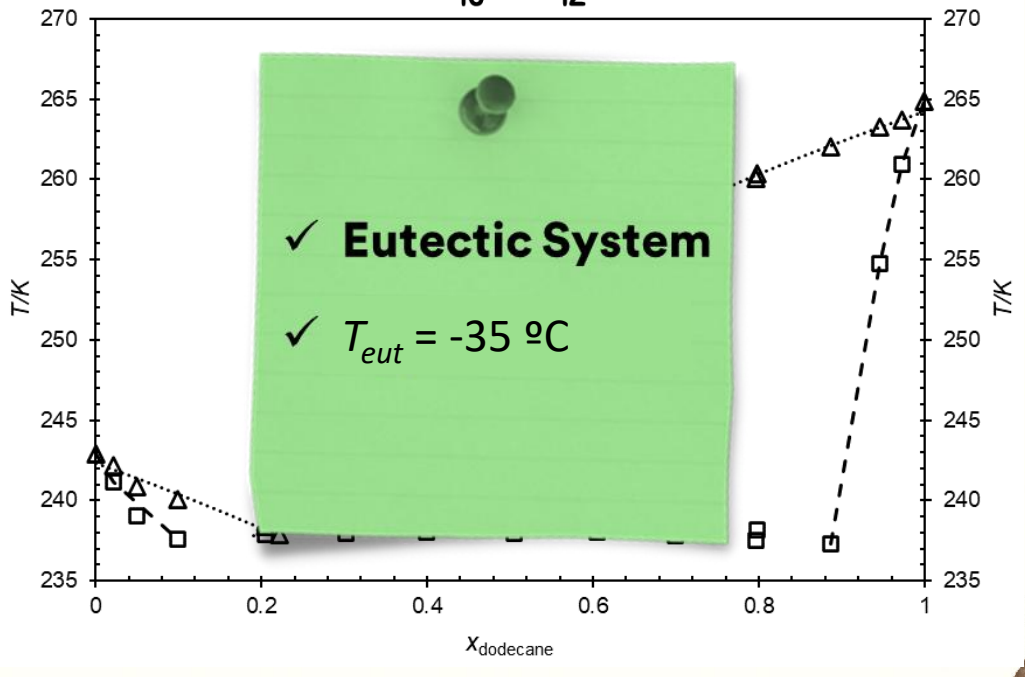


Fig. 10 - Binary solid-liquid phase diagram of  $n\text{-C}_{10}$  and  $n\text{-C}_{11}$ .

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# Binary Phase Diagrams

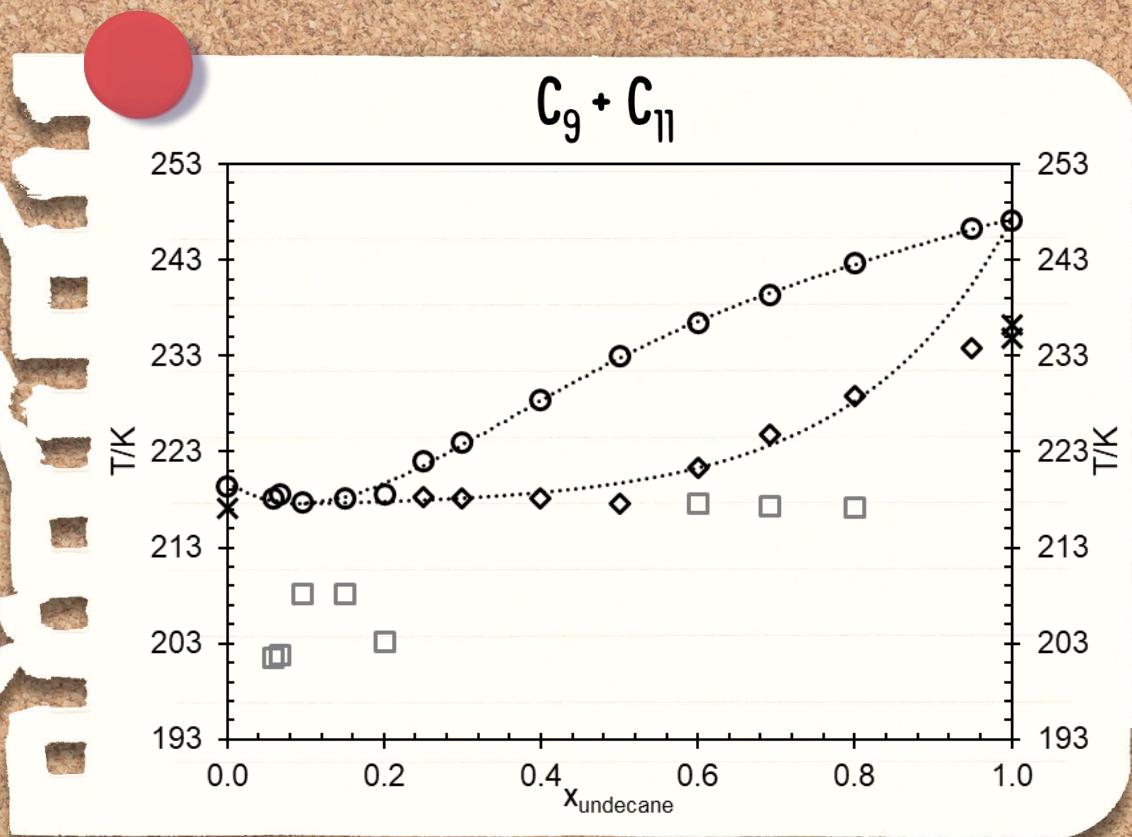


Fig. 11 – Binary solid-liquid phase diagram of  $n-C_9$  and  $n-C_{11}$  (submitted to the International Journal of Thermophysics).

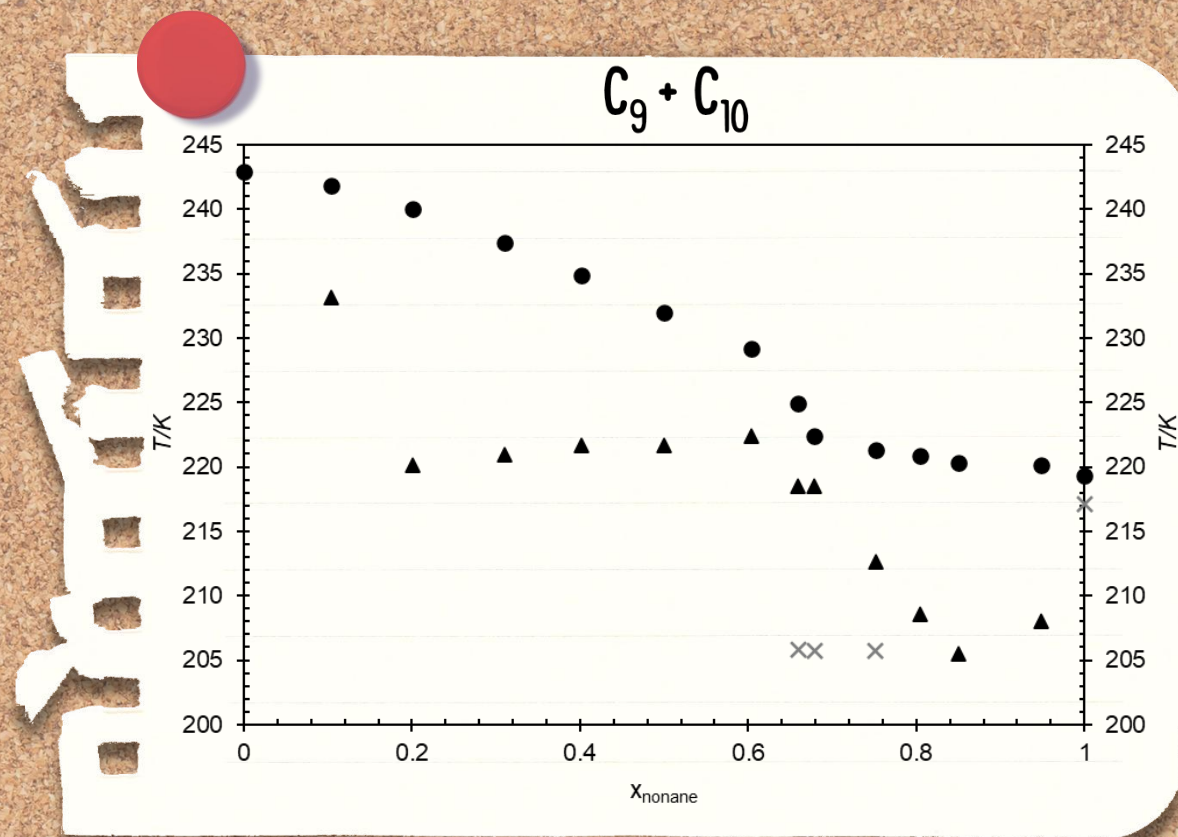


Fig. 12 – Preliminary binary solid-liquid phase diagram of  $n-C_9$  and  $n-C_{10}$ .

# Binary Phase Diagrams

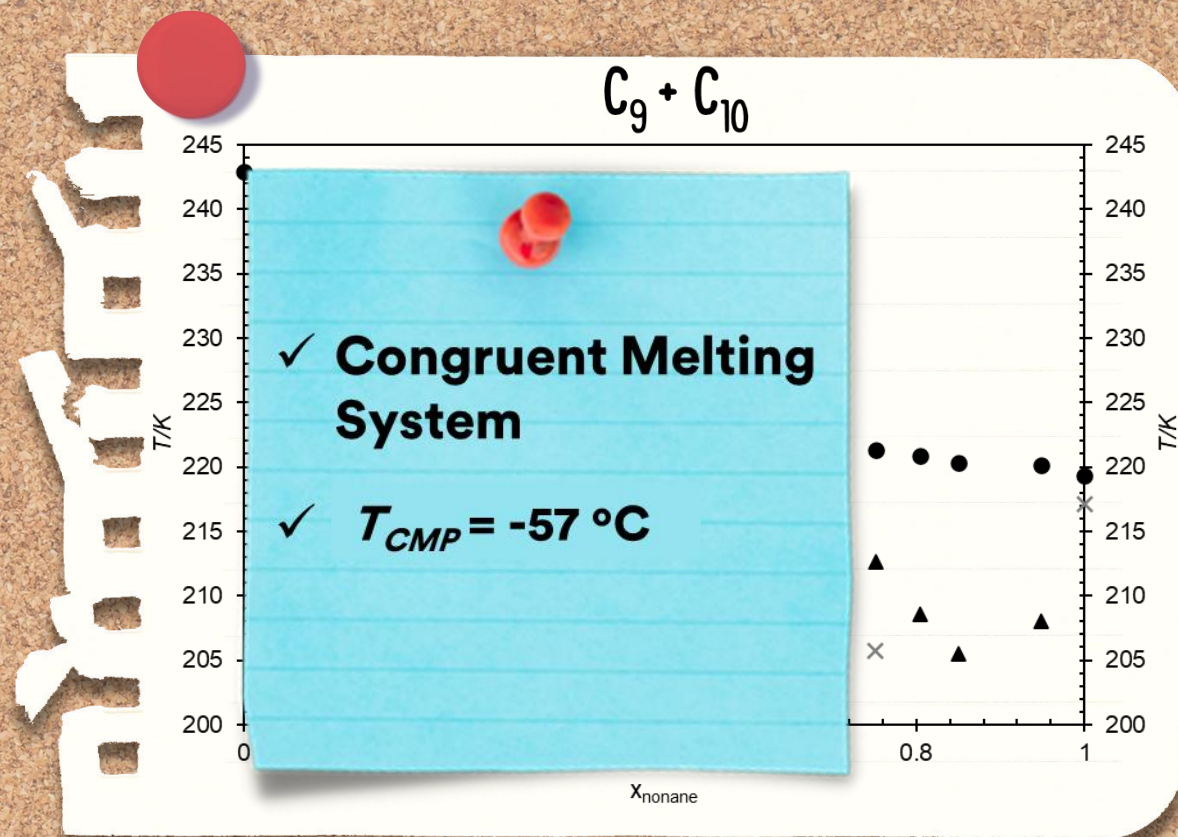
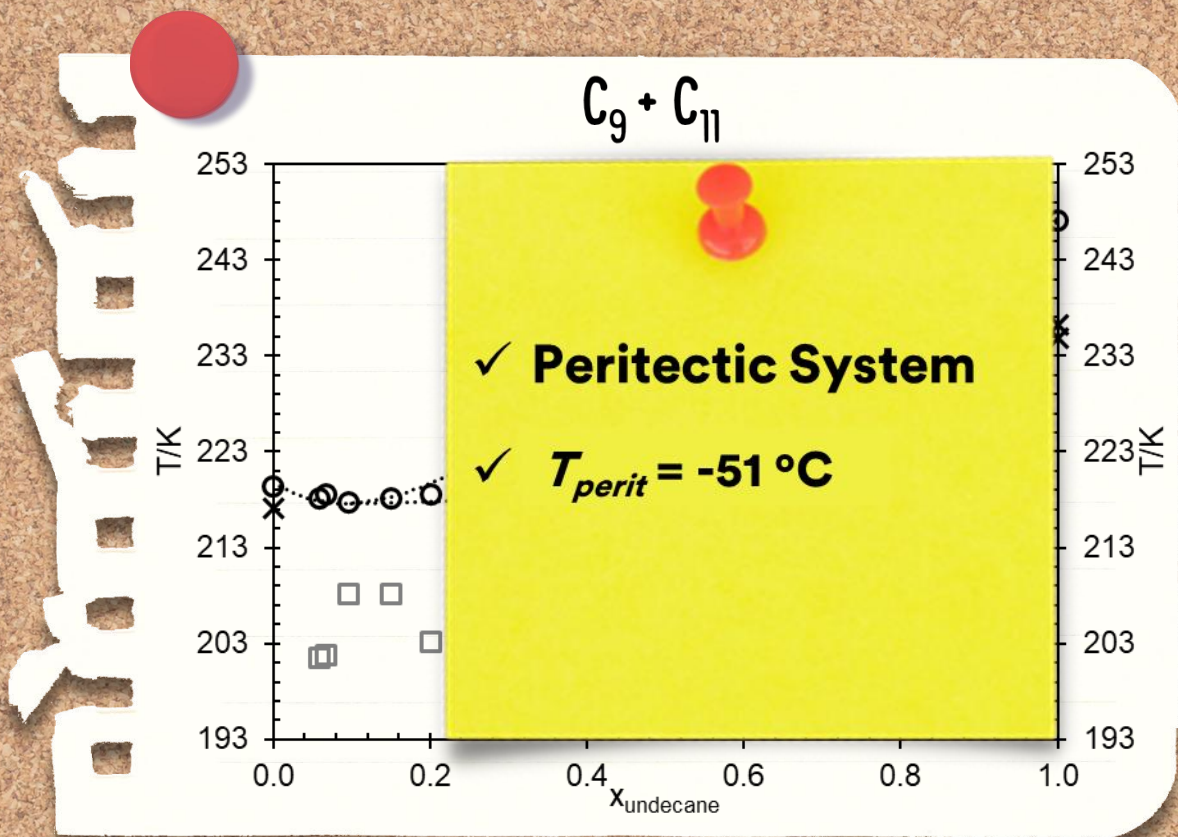


Fig. 11 – Binary solid-liquid phase diagram of  $n-C_9$  and  $n-C_{11}$  (submitted to the International Journal of Thermophysics).

Fig. 12 – Preliminary binary solid-liquid phase diagram of  $n-C_9$  and  $n-C_{10}$ .

Odd-even  
effects

Grouping of  
characteristics

*n* - Alkanes

Chain size

Polymorphism

Phase  
Equilibrium

Predict the  
potential as PCM<sub>s</sub>

# Conclusions

Odd-Even effects  
identified and  
analyzed

Phase equilibrium  
studies for odd and  
even alkanes

3/4 alkane systems  
with good potential as  
PCMs for low T

Prediction of the  
phase equilibrium  
characteristics

Fulfill the lack of  
available data on  
low T PCM systems

# Acknowledgements

This work was supported by Fundação para a Ciência e a Tecnologia (FCT), Portugal, Projects UIDB/00100/2020 (<https://doi.org/10.54499/UIDB/00100/2020>), UIDP/00100/2020 (<https://doi.org/10.54499/UIDP/00100/2020>), and IMS-LA/P/0056/2020 UIDB/00100/2020 (<https://doi.org/10.54499/LA/P/0056/2020>).

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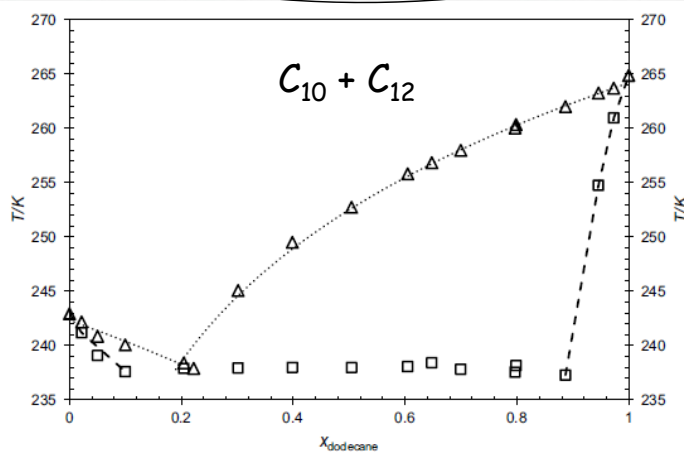


# Literature vs Our Results

Even + Even

↓  
Eutectic

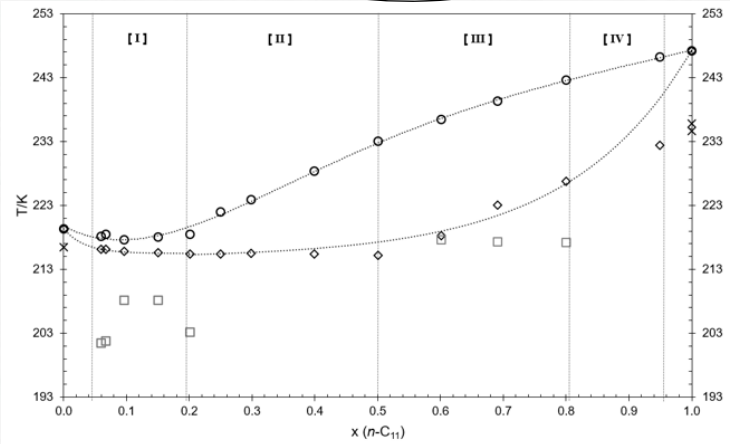
$C_8 + C_{10} / C_{10} + C_{12}$



Odd + Odd

↓  
Congruent  
Melting

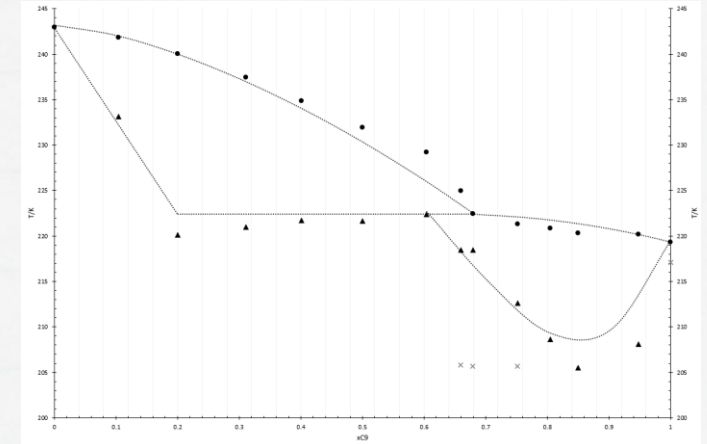
$C_9 + C_{11}$



Even + Odd

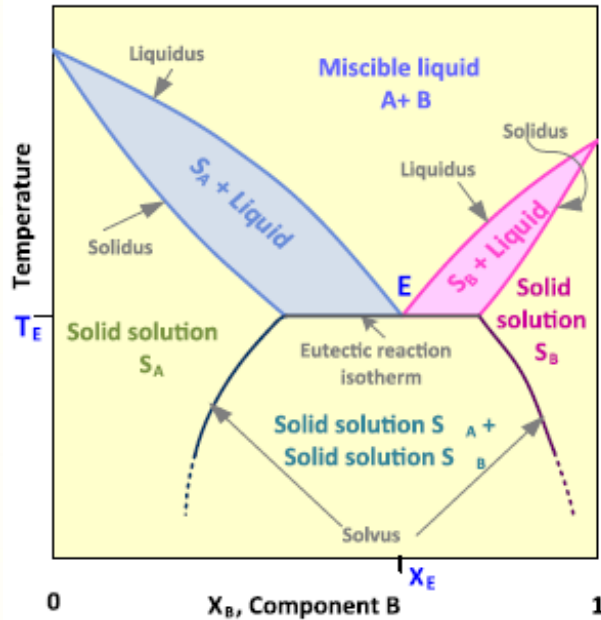
↓  
Peritectic

$C_9 + C_{10}$

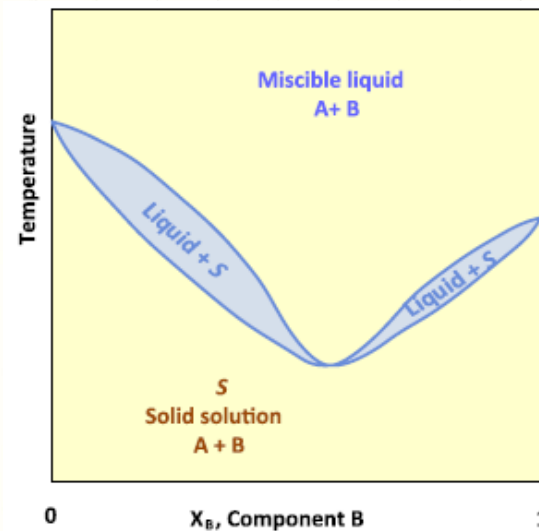


# Phase Equilibrium in TES

## Eutectic Systems



## Congruent Melting Systems



## Peritectic Systems

